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and a salt of that weak acid or

fully in aqueous solution a

base no combining a strong

solution of a weak acid in water

acid and a strong base will

is a mixture of the nonionized

produce salt and water excess

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not act as a buffer

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because it can deprotonate
alkynes alcohols and a host of
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neutralize a weak acid with a

strong base we get a salt that

contains the conjugate base of

the weak acid this conjugate

base is usually a weak base for

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reaction of the weak acid acetic

acid with the strong base

sodium hydroxide

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