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photograph of a
separatory funnel in a
laboratory scale
extraction of 2
immiscible liquids
liquids are a diethyl
ether upper phase and a
lower aqueous phase
soxhlet extractor
extraction in chemistry
is a separation process
consisting of the
separation of a
substance from a matrix
the distribution
liquid liquid extraction

chemical engineering

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liquid extraction llx is

a mass transfer

operation which is used

to separate components

of a liquid mixture by

contacting it with

another liquid the feed

stream may consist of

any number of components

it must have at least

two components

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extraction of neutral

compounds if the desired

organic compound is

neutral i e is neither

acidic nor basic the

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7 6 solvent partitioning

liquid liquid extraction

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university this method

is often called

extraction extraction

means drawing a compound

out of a mixture using a

solvent solvent

partitioning is more

specific it means

compounds have a choice

of two solvents that

they can dissolve in

some compounds dissolve

in one solvent

homogeneous liquid

liquid extraction as an

alternative sample - Aug

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liquid extraction is a
widely used technique of
sample preparation in
biomedical analysis in
spite of the high pre
concentration capacities
of liquid liquid
extraction it suffers
from a number of
limitations including
time and effort
consumption large
organic solvent
utilization and poor
performance in highly
polar analytes
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extraction technology
exploits the relative
ability of solutes to
distribute between
immiscible liquid phases
in contact to recover
target components from
primary feed mixtures
theoretical strategies
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liquid liquid extraction
systems model aspects of
the problem relevant to
phase equilibria
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extractions are
versatile purification

technologies which
involve the mass
transfer of several
compounds from one phase
to another through
partially miscible
liquid phases that
interact with each other
during their intimate
contact from
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liquid extraction with
solvents more dense than
water liquid liquid
extraction is a method

by which a compound is pulled from solvent a to solvent b where solvents a and b are not miscible the most common method of liquid liquid extraction is performed using a separatory funnel

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extraction is a countercurrent separation process for isolating the constituents of a liquid mixture in its simplest form this involves the extraction of a solute from a binary solution by bringing it into contact with a second immiscible solvent in which the solute is soluble

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an overview

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extraction is an
important separation
method in research and

chemical analysis as a commercial process it is frequently used in the chemical and mining industries and in the downstream recovery of fermentation products antibiotics amino acids steroids

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it is most common to use
liquid liquid extraction
a process that occurs in
a separatory funnel a
solution containing
dissolved components is
placed in the funnel and
an immiscible solvent is
added resulting in two
layers that are shaken

together
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liquid liquid extraction
method has been
developed to produce
liquid pcs via simple
mixing and liquid phase
separation at room
temperature single and
two component
extractants are used to
extract part of the
solvent from a diluted
colloidal solution which
effectively concentrates
the colloidal solution
without
liquid liquid extraction

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extraction lle also
known as solvent
extraction and
partitioning is a method
to separate compounds or
metal complexes based on
their relative
solubilities in two
different immiscible
liquids usually water
polar and an organic
solvent non polar
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a solid liquid
extraction technique
capable of extracting a

wide range of organic
contaminants from a wide
variety of biological
matrices 6 ple offers
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efficiency sample
throughput and
automation 7 ple can be
viewed as an extension
of supercritical fluid
extraction
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extraction lle also
known as solvent
extraction is a typical
ternary system in the
chemical engineering
field which separates

chemicals from one
solution to another
based on the different
solubility of the solute
chemical in
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