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purpose what is an isotope what does
it mean to say that the atoms in a
sample of an element are isotopes of
each other ordinary beans are a lot
bigger than atoms but perhaps they
can give you one or two clues about

isotopes we have talked in class about isotopes atoms of the same element that differ in [beanium lab](#) - Aug 03 2022 web beanium isotope lab section 1 pre lab questions define average atomic mass write a mathematical equation that shows how you would determine the average atomic mass of an element section 2 data table don't forget calibration and units total all beans 100 beanium isotope blackium beanium isotope brownium beanium

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lentilium and then mass each group
record the values in the data table
2 count the number of each isotope
by counting the pintonium navium and
lentilium 3 calculate the total
number of isotopes

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classroom mixture of beanium sort
your beanium into its three isotopic
bean types count the number of beans
in each pile find the mass of each
pile of beans determine the average
mass of each type of bean based on
the samples masses

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the beans 2 do not throw the beans 3
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