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behaviour was made by antoine laurent lavoisier at the end of the 18th century lavoisier supposed that all acids must contain oxygen and this idea was incorporated in the names used for this element in the various languages the english

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equilibrium formulated in scheme 1 between hydrogen ion and the corresponding acid and base can be called a simple acid base equilibrium by mixing two simple systems a double acid base system and an acid base equilibrium result that can always be formulated as follows acid1 base2 acid2 base1

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discovery by 1890 the swedish chemist svante arrhenius 1859 1927 was able to formulate the first useful theory of acids arrhenius definition an acidic substance is one whose molecular unit contains at least one

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across different contexts one such foundational principle is the acid base behavior of molecules in the general chemistry sequence the brønsted lowry theory is stressed because it lends itself well to studying equilibrium and kine

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