

**WJ Hussar**

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 web a substance may be assigned to one our four conceivable categories it may be an acid or a base but in addition it may be both an acid and a base or it may be neither an acid nor  
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 most of the acids are solid it has a

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 s14 3 21 strong electrolytes are 100 ionized and as long as the component ions are neither weak acids nor weak bases the ionic species present result

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acids  $\text{CH}_3\text{COOH}$   $\text{CO}_2$

$\text{HNO}_2$   $\text{HCOOH}$   $\text{CCl}_4$   $\text{CH}_3\text{COOH}$

$\text{HNO}_3$   $\text{H}_2\text{CO}_3$   $\text{HCl}$   $\text{H}_2\text{O}$

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which of the

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