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nuclear symbols mass number p no 235

92 u atomic number number of p  
element symbol types of radioactive  
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nucleus 238 4 234 92 u 2 he 90th 0  
beta production b 1 e 234 234 90th  
91pa 0 1e 4 2 he 2 alpha radiation  
limited to very large nuclei

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chemistry the spontaneous  
disintegration of a nucleus into a  
slightly lighter and more stable  
nucleus accompanied by emission of  
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the mass defect for the formation of  
an oxygen 16 nucleus in both grams  
and g mol and calculate the binding  
energy in both mev nucleon and kj mol  
the mass of an 16 o atom is 15 99491  
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transmutation reactions in a nuclear decay reaction also called radioactive decay an unstable nucleus emits radiation and is transformed into the nucleus of one or more other elements

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 radiation equal to the amount of  
 radiation that produces  $2 \times 10^9$  ion  
 pairs when it passes through 1 cm<sup>3</sup>  
 of dry air rem the quantity of  
 ionizing radiation that does as much  
 damage to human tissue as

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web nuclide identified by the number  
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nucleus mass defect difference  
 between the mass of an atom and the  
 sum of the masses of its protons  
 neutrons and electrons binding energy  
 per nucleon binding energy of the  
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