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chemical reaction combustion
reaction a chemical reaction that
occurs when a substance reacts
with oxygen releasing energy in
the form of heat and light
decomposition reaction a chemical
reaction that occurs when a
single compound breaks down
into two or more elements or
new compounds

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number that appears before a
formula in an equation spectator

ion a particle not directly
involved in a chemical reaction
combustion reaction a reaction in
which oxygen reacts with
another substance often
producing light or heat reactant a
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only involve electrons positions

in the developing and breaking

of chemical bonds among atoms

with no change to the nuclei this

quiz will give you a positive

reaction of

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reactions in the space provided

write the letter of the term or

phrase that best completes each

statement or best answers each

question 1 you mix solution a

with solution b in a beaker which

of the following observations does

not help you prove that a

chemical reaction has occurred

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chemical equation that does not
indicate relative amounts of
reactants and products skeleton
equation a new substance formed
in a chemical reaction product a
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terms in this set 20 the
production of one or more new
substances the only sure evidence
for a chemical reaction is inhibitor
a material used to decrease the
rate of a chemical reaction is a n
matter is not created or destroyed

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reactions are part of our daily
lives from cooking in the kitchen
to driving a car these reactions
are commonplace in a chemical
reaction the molecules of one
substance break apart and join
together with those of another

substance to create a different
compound combination of
molecules many chemical
reactions are non

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